

FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: Pharmacology

SEMESTER: II CODE: BP201T

NAME: Human Anatomy and Physiology II -Theory

Teaching & Evaluation Scheme:-

		Teac	hing So	cheme	(Hours)					Evalua	ation Sche	me			
Subject	Name of the					Credits		The	eory			Prac	ctical		
Code	Subject	Th	Tu	Pr	Total		Intern Exan		End Seme Exam		Interi Exai		End Sem Exan		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
	Human Anatomy	2		0	4	4	15	1	7.5	2					100
BP201T	and Physiology II -Theory	3	1	0	4	4	10 (CM)		75	3					100

Scope: This subject is designed to impart fundamental knowledge on the structure and functions of the various systems of the human body. It also helps in understanding both homeostatic mechanisms. The subject provides the basic knowledge required to understand the various disciplines of pharmacy.

Objectives: Upon completion of this course the student should be able to:

- Explain the gross morphology, structure and functions of various organs of the human body.
- Describe the various homeostatic mechanisms and their imbalances.
- Identify the various tissues and organs of different systems of human body.
- Perform the hematological tests like blood cell counts, haemoglobin estimation, bleeding/clotting time etc and also record blood pressure, heart rate, pulse and respiratory volume.
- Appreciate coordinated working pattern of different organs of each system
- Appreciate the interlinked mechanisms in the maintenance of normal functioning (homeostasis) of human body.



COURSE CONTENT (45 Hour)

UNIT-I 10 Hours

Nervous system

Organization of nervous system, neuron, neuroglia, classification and properties of nerve fibre, electrophysiology, action potential, nerve impulse, receptors, synapse, neurotransmitters.

Central nervous system: Meninges, ventricles of brain and cerebrospinal fluid.structure and functions of brain (cerebrum, brain stem, cerebellum), spinal cord (gross structure, functions of afferent and efferent nerve tracts, reflex activity)

UNIT-II 06 Hours

• Digestive system

Anatomy of GI Tract with special reference to anatomy and functions of stomach, (Acid production in the stomach, regulation of acid production through parasympathetic nervous system, pepsin role in protein digestion) small intestine and large intestine, anatomy and functions of salivary glands, pancreas and liver, movements of GIT, digestion and absorption of nutrients and disorders of GIT.

Energetics

Formation and role of ATP, Creatinine Phosphate and BMR.

UNIT-III 10 Hours

• Respiratory system

Anatomy of respiratory system with special reference to anatomy of lungs, mechanism of respiration, regulation of respiration

Lung Volumes and capacities transport of respiratory gases, artificial respiration, and resuscitation methods.

Urinary system

Anatomy of urinary tract with special reference to anatomy of kidney and nephrons, functions of kidney and urinary tract, physiology of urine formation, micturition reflex and role of kidneys in acid base balance, role of RAS in kidney and disorders of kidney.



UNIT-IV 10 Hours

• Endocrine system

Classification of hormones, mechanism of hormone action, structure and functions of pituitary gland, thyroid gland, parathyroid gland, adrenal gland, pancreas, pineal gland, thymus and their disorders.

UNIT-V 09 Hours

• Reproductive system

Anatomy of male and female reproductive system, Functions of male and female reproductive system, sex hormones, physiology of menstruation, fertilization, spermatogenesis, oogenesis, pregnancy and parturition

• Introduction to genetics

Chromosomes, genes and DNA, protein synthesis, genetic pattern of inheritance



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: Pharmacology

SEMESTER: II CODE: BP207P

NAME: Human Anatomy and Physiology II - Practical

Teaching & Evaluation Scheme:-

		Teac	hing S	chem	e (Hours)					Eval	uation Sc	heme			
Subject	Name of the					Credits		Th	eory			Prac	ctical		
Code	Subject	Th	Tu	Pr	Total			Internal End Semeste Exam Exam			Interi Exai		End Sem Exan		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
D D207D	Human Anatomy				4	_					10	4	2.5		-0
BP207P	and Physiology II - Practical			4	4	2					5 (CM)		35	4	50

Practical physiology is complimentary to the theoretical discussions in physiology. Practicals allow the verification of physiological processes discussed in theory classes through experiments on living tissue, intact animals or normal human beings. This is helpful for developing an insight on the subject.

- 1. To study the integumentary and special senses using specimen, models, etc.,
- 2. To study the nervous system using specimen, models, etc.,
- 3. To study the endocrine system using specimen, models, etc
- 4. To demonstrate the general neurological examination
- 5. To demonstrate the function of olfactory nerve
- 6. To examine the different types of taste.
- 7. To demonstrate the visual acuity
- 8. To demonstrate the reflex activity
- 9. Recording of body temperature
- 10. To demonstrate positive and negative feedback mechanism.
- 11. Determination of tidal volume and vital capacity.
- 12. Study of digestive, respiratory, cardiovascular systems, urinary and reproductive systems with the help of models, charts and specimens.
- 13. Recording of basal mass index
- 14. Study of family planning devices and pregnancy diagnosis test.
- 15. Demonstration of total blood count by cell analyser
- 16. Permanent slides of vital organs and gonads.



Recommended Books (Latest Editions)

- 1. Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee brothers medical publishers, New Delhi.
- 2. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York
- 3. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co, Riverview, MI USA
- 4. Text book of Medical Physiology- Arthur C,Guyton and John. E. Hall. Miamisburg, OH, U.S.A.
- 5. Principles of Anatomy and Physiology by Tortora Grabowski. Palmetto, GA, U.S.A.
- 6. Textbook of Human Histology by Inderbir Singh, Jaypee brothers medical publishers, New Delhi.
- 7. Textbook of Practical Physiology by C.L. Ghai, Jaypee brothers medical publishers, New Delhi.
- 8. Practical workbook of Human Physiology by K. Srinageswari and Rajeev Sharma, Jaypee brother's medical publishers, New Delhi.

Reference Books:

- Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co, Riverview, MI USA
- 2. Text book of Medical Physiology- Arthur C, Guyton and John. E. Hall. Miamisburg, OH, U.S.A.
- 3. Human Physiology (vol 1 and 2) by Dr. C.C. Chatterrje, Academic Publishers Kolkata.



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: Pharmaceutical Chemistry & Pharmaceutical Analysis

SEMESTER: II CODE: BP202T

NAME: Pharmaceutical Organic Chemistry I - Theory

Teaching & Evaluation Scheme:-

		Teac	hing S	cheme	e (Hours)					Evalua	tion Sche	me			
Subject	Name of the					Credits		The	eory			Prac	ctical		
Code	Subject	Th	Tu	Pr	Total		Internal	Exam	End Semo Exam		Interi Exai		End Sem Exar		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
ррасат	Pharmaceutical	_			_	_	15	1	7.5	_					100
BP202T C	Organic Chemistry I -Theory	3	I	0	4	4	10 (CM)		75	3					100

Scope: This subject deals with classification and nomenclature of simple organic compounds, structural isomerism, intermediates forming in reactions, important physical properties, reactions and methods of preparation of these compounds. The syllabus also emphasizes on mechanisms and orientation of reactions.

Objectives: Upon completion of the course the student shall be able to

- write the structure, name and the type of isomerism of the organic compound
- write the reaction, name the reaction and orientation of reactions
- account for reactivity/stability of compounds,
- identify/confirm the identification of organic compound

COURSE CONTENT (45 Hour)

General methods of preparation and reactions of compounds superscripted with asterisk (*) to be explained

To emphasize on definition, types, classification, principles/mechanisms, applications, examples and differences



UNIT-I 07 Hours

• Classification, nomenclature and isomerism

Classification of Organic Compounds Common and IUPAC systems of nomenclature of organic compounds (up to 10 Carbons open chain and carbocyclic compounds)

Structural isomerisms in organic compounds.

UNIT-II 10 Hours

• Alkanes*, Alkenes* and Conjugated dienes*

SP³ hybridization in alkanes, Halogenation of alkanes, uses of paraffins. Stabilities of alkenes, SP² hybridization in alkenes

 E_1 and E_2 reactions – kinetics, order of reactivity of alkyl halides, rearrangement of carbocations, Saytzeffs orientation and evidences. E_1 verses E_2 reactions, Factors affecting E_1 and E_2 reactions. Ozonolysis, electrophilic addition reactions of alkenes, Markownikoff's orientation, free radical addition reactions of alkenes, Anti Markownikoff's orientation.

Stability of conjugated dienes, Diel-Alder, electrophilic addition, free radical addition reactions of conjugated dienes, allylic rearrangement.

UNIT-III 10 Hours

Alkyl halides*

 SN_1 and SN_2 reactions - kinetics, order of reactivity of alkyl halides, stereochemistry and rearrangement of carbocations.

 SN_1 versus SN_2 reactions, Factors affecting SN_1 and SN_2 reactions

Structure and uses of ethylchloride, Chloroform, trichloroethylene, tetrachloroethylene, dichloromethane, tetrachloromethane and iodoform.

 Alcohols*- Qualitative tests, Structure and uses of Ethyl alcohol, Methyl alcohol, chlorobutanol, Cetosteryl alcohol, Benzyl alcohol, Glycerol, Propylene glycol.



UNIT-IV 10 Hours

• Carbonyl compounds* (Aldehydes and ketones)

Nucleophilic addition, Electromeric effect, aldol condensation, Crossed Aldol condensation, Cannizzaro reaction, Crossed Cannizzaro reaction, Benzoin condensation, Perkin condensation, qualitative tests, Structure and uses of Formaldehyde, Paraldehyde, Acetone, Chloral hydrate, Hexamine, Benzaldehyde, Vanilin, Cinnamaldehyde.

UNIT-V 08 Hours

• Carboxylic acids*

Acidity of carboxylic acids, effect of substituents on acidity, inductive effect and qualitative tests for carboxylic acids, amide and ester

Structure and Uses of Acetic acid, Lactic acid, Tartaric acid, Citric acid, Succinic acid. Oxalic acid, Salicylic acid, Benzoic acid, Benzyl benzoate, Dimethyl phthalate, Methyl salicylate and Acetyl salicylic acid

• Aliphatic amines* - Basicity, effect of substituent on Basicity. Qualitative test, Structure and uses of Ethanolamine, Ethylenediamine, Amphetamine.



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: Pharmaceutical Chemistry & Pharmaceutical Analysis

SEMESTER: II CODE: BP208P

NAME: Pharmaceutical Organic Chemistry I - Practical

Teaching & Evaluation Scheme:-

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		Teac	hing S	chem	e (Hours)					Eval	uation Sc	heme			
Subject	Name of the Subject					Credits		Th	eory			Prac	ctical		
Code	Code Name of the Subject		Tu	Pr	Total		Interi Exai		End Semo		Interi Exar		End Sem Exan		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
D D200D	Pharmaceutical			4	4	2					10	4	25	4	5 0
BP208P	Organic Chemistry I - Practical			4	4	2					5 (CM)		35	4	50

- I. Systematic qualitative analysis of unknown organic compounds like
 - 1. Preliminary test: Color, odour, aliphatic/aromatic compounds, saturation and unsaturation, etc.
 - 2. Detection of elements like Nitrogen, Sulphur and Halogen by Lassaigne's test
 - 3. Solubility test
 - 4. Functional group test like Phenols, Amides/ Urea, Carbohydrates, Amines, Carboxylic acids, Aldehydes and Ketones, Alcohols, Esters, Aromatic and Halogenated Hydrocarbons, Nitro compounds and Anilides.
 - 5. Melting point/Boiling point of organic compounds
 - 6. Identification of the unknown compound from the literature using melting point/ boiling point.
 - 7. Preparation of the derivatives and confirmation of the unknown compound by melting point/ boiling point.
 - 8. Minimum 5 unknown organic compounds to be analysed systematically.
- II. Preparation of suitable solid derivatives from organic compounds.
- III. Construction of molecular models



Recommended Books (Latest Editions)

- 1. Organic Chemistry by Morrison and Boyd
- 2. Organic Chemistry by I.L. Finar, Volume-I
- 3. Textbook of Organic Chemistry by B.S. Bahl & Arun Bahl.
- 4. Organic Chemistry by P.L.Soni
- 5. Practical Organic Chemistry by Mann and Saunders.
- 6. Vogel's text book of Practical Organic Chemistry
- 7. Advanced Practical organic chemistry by N.K.Vishnoi.
- 8. Introduction to Organic Laboratory techniques by Pavia, Lampman and Kriz.
- 9. Reaction and reaction mechanism by Ahluwaliah/Chatwal.



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: Pharmaceutical Chemistry & Pharmaceutical Analysis

SEMESTER: II CODE: BP203T

NAME: Biochemistry –Theory

Teaching & Evaluation Scheme:-

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		Teac	hing S	cheme	e (Hours)					Evalua	ation Sche	me			
Subject	Name of the					Credits		The	eory			Prac	ctical		
Code	Subject	Th	Tu Pr To		Total		Intern Exan		End Seme Exam		Interi Exai		End Sem Exan		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
DD202T	Biochemistry	2	1	0	4	4	15	1	75	2					100
BP2031	P203T Hochemstry -Theory	3	1	U	4	4	10 (CM)		/3	3					100

Scope: Biochemistry deals with complete understanding of the molecular levels of the chemical process associated with living cells. The scope of the subject is providing biochemical facts and the principles to understand metabolism of nutrient molecules in physiological and pathological conditions. It is also emphasizing on genetic organization of mammalian genome and hetero & autocatalytic functions of DNA.

Objectives: Upon completion of course student shell able to

- Understand the catalytic role of enzymes, importance of enzyme inhibitors in design of new drugs, therapeutic and diagnostic applications of enzymes.
- Understand the metabolism of nutrient molecules in physiological and pathological conditions.
- Understand the genetic organization of mammalian genome and functions of DNA in the synthesis of RNAs and proteins.

COURSE CONTENT (45 Hour)

UNIT-I 08 Hours

Biomolecules

Introduction, classification, chemical nature and biological role of carbohydrate, lipids, nucleic acids, amino acids and proteins.



• Bioenergetics

Concept of free energy, endergonic and exergonic reaction, Relationship between free energy, enthalpy and entropy; Redox potential.

Energy rich compounds; classification; biological significances of ATP and cyclic AMP

UNIT-II 10 Hours

• Carbohydrate metabolism

Glycolysis – Pathway, energetics and significance Citric acid cycle- Pathway, energetics and significance

HMP shunt and its significance; Glucose-6-Phosphate dehydrogenase (G6PD) deficiency Glycogen metabolism Pathways and glycogen storage diseases (GSD) Gluconeogenesis-Pathway and its significance

Hormonal regulation of blood glucose level and Diabetes mellitus

• Biological oxidation

Electron transport chain (ETC) and its mechanism.

Oxidative phosphorylation & its mechanism and substrate phosphorylation

Inhibitors ETC and oxidative phosphorylation/Uncouplers

UNIT-III 10 Hours

Lipid metabolism

β-Oxidation of saturated fatty acid (Palmitic acid)

Formation and utilization of ketone bodies; ketoacidosis De novo synthesis of fatty acids (Palmitic acid)

Biological significance of cholesterol and conversion of cholesterol into bile acids, steroid hormone and vitamin D

Disorders of lipid metabolism: Hypercholesterolemia, atherosclerosis, fatty liver and obesity.



Amino acid metabolism

General reactions of amino acid metabolism: Transamination, deamination & decarboxylation, urea cycle and its disorders

Catabolism of phenylalanine and tyrosine and their metabolic disorders (Phenyketonuria,

Albinism, alkeptonuria, tyrosinemia)

Synthesis and significance of biological substances; 5-HT, melatonin, dopamine, noradrenaline, adrenaline

Catabolism of heme; hyperbilirubinemia and jaundice.

UNIT-IV 10 Hours

• Nucleic acid metabolism and genetic information transfer

Biosynthesis of purine and pyrimidine nucleotides

Catabolism of purine nucleotides and Hyperuricemia and Gout disease

Organization of mammalian genome

Structure of DNA and RNA and their functions

DNA replication (semi conservative model)

Transcription or RNA synthesis

Genetic code, Translation or Protein synthesis and inhibitors

UNIT-V 07 Hours

Enzymes

Introduction, properties, nomenclature and IUB classification of enzymes Enzyme kinetics (Michaelis plot, Line Weaver Burke plot)

Enzyme inhibitors with examples

Regulation of enzymes: enzyme induction and repression, allosteric enzymes regulation

Therapeutic and diagnostic applications of enzymes and isoenzymes Coenzymes -

Structure and biochemical functions.



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: Pharmaceutical Chemistry & Pharmaceutical Analysis

SEMESTER: II CODE: BP209P

NAME: Biochemistry - Practical

Teaching & Evaluation Scheme:-

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		Teac	hing S	chemo	e (Hours)					Eval	uation Sc	heme			
Subject	Name of the					Credits		Th	eory			Prac	ctical		
Code	Subject	Th	Tu	Pr	Total		Interi Exai		End Seme Exam		Interi Exai		End Sem Exan		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
BP209P	Biochemistry -			4	4	2					10	4	35	1	50
D1 2091	Practical			+	_						5 (CM)		33	-	30

- 1. Qualitative analysis of carbohydrates (Glucose, Fructose, Lactose, Maltose, Sucrose and starch)
- 2. Identification tests for Proteins (albumin and Casein)
- 3. Quantitative analysis of reducing sugars (DNSA method) and Proteins (Biuret method)
- 4. Qualitative analysis of urine for abnormal constituents
- 5. Determination of blood creatinine
- 6. Determination of blood sugar
- 7. Determination of serum total cholesterol
- 8. Preparation of buffer solution and measurement of pH
- 9. Study of enzymatic hydrolysis of starch
- 10. Determination of Salivary amylase activity
- 11. Study the effect of Temperature on Salivary amylase activity.
- 12. Study the effect of substrate concentration on salivary amylase activity.



Recommended Books (Latest Editions)

- 1. Principles of Biochemistry by Lehninger.
- 2. Harper's Biochemistry by Robert K. Murry, Daryl K. Granner and Victor W. Rodwell.
- 3. Biochemistry by Stryer.
- 4. Biochemistry by D. Satyanarayan and U.Chakrapani
- 5. Textbook of Biochemistry by Rama Rao.
- 6. Textbook of Biochemistry by Deb.
- 7. Outlines of Biochemistry by Conn and Stumpf
- 8. Practical Biochemistry by R.C. Gupta and S. Bhargavan.
- 9. Introduction of Practical Biochemistry by David T. Plummer. (3rd Edition)
- 10. Practical Biochemistry for Medical students by Rajagopal and Ramakrishna.
- 11. Practical Biochemistry by Harold Varley.



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: Pharmacology

SEMESTER: II CODE: BP204T

NAME: Pathophysiology -Theory

Teaching & Evaluation Scheme:-

		Teac	hing So	cheme	(Hours)					Evalua	ation Sche	me			
Subject	ĭ1				Credits		The	eory			Prac	ctical			
Code	Subject	Th	Tu	Pr	Total		Internal End Semester Exam Exam				Interi Exai		End Sem Exan		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
BP204T	Pathophysiology -Theory	3	1	0	4	4	15 10 (CM)	1	75	3	1		1	1	100

Scope: Pathophysiology is the study of causes of diseases and reactions of the body to such disease producing causes. This course is designed to impart a thorough knowledge of the relevant aspects of pathology of various conditions with reference to its pharmacological applications, and understanding of basic pathophysiological mechanisms. Hence it will not only help to study the syllabus of pathology, but also to get baseline knowledge required to practice medicine safely, confidently, rationally and effectively.

Objectives: Upon completion of the subject student shall be able to –

- Describe the etiology and pathogenesis of the selected disease states;
- Name the signs and symptoms of the diseases; and
- Mention the complications of the diseases.

COURSE CONTENT (45 Hour)

UNIT-I 10 Hours

• Basic principles of Cell injury and Adaptation:

Introduction, definitions, Homeostasis, Components and Types of Feedback systems, Causes of cellular injury, Pathogenesis (Cell membrane damage, Mitochondrial damage, Ribosome damage, Nuclear damage), Morphology of cell injury – Adaptive changes



(Atrophy, Hypertrophy, hyperplasia, Metaplasia, Dysplasia), Cell swelling, Intra cellular accumulation, Calcification, Enzyme leakage and Cell Death Acidosis & Alkalosis, Electrolyte imbalance

Basic mechanism involved in the process of inflammation and repair:

Introduction, Clinical signs of inflammation, Different types of Inflammation, Mechanism of Inflammation – Alteration in vascular permeability and blood flow, migration of WBC's, Mediators of inflammation, Basic principles of wound healing in the skin, Pathophysiology of Atherosclerosis

UNIT-II 10 Hours

- Cardiovascular System:
 - Hypertension, congestive heart failure, ischemic heart disease (angina, myocardial infarction, atherosclerosis and arteriosclerosis)
- **Respiratory system:** Asthma, Chronic obstructive airways diseases.
- Renal system: Acute and chronic renal failure

UNIT-III 10 Hours

• Haematological Diseases:

Iron deficiency, megaloblastic anemia (Vit B_{12} and folic acid), sickle cell anemia, thalasemia, hereditary acquired anemia, hemophilia

- Endocrine system: Diabetes, thyroid diseases, disorders of sex hormones
- **Nervous system:** Epilepsy, Parkinson's disease, stroke, psychiatric disorders: depression, schizophrenia and Alzheimer's disease.
- Gastrointestinal system: Peptic Ulcer

UNIT-IV 08 Hour

- Inflammatory bowel diseases, jaundice, hepatitis (A, B, C, D, E, F) alcoholic liver disease.
- **Disease of bones and joints:** Rheumatoid arthritis, osteoporosis and gout
- **Principles of cancer:** classification, etiology and pathogenesis of cancer
- **Diseases of bones and joints:** Rheumatoid Arthritis, Osteoporosis, Gout
- **Principles of Cancer:** Classification, etiology and pathogenesis of Cancer



UNIT-V 07 Hour

- Infectious diseases: Meningitis, Typhoid, Leprosy, Tuberculosis, Urinary tract infections
- Sexually transmitted diseases: AIDS, Syphilis, and Gonorrhea

Recommended Books (Latest Editions)

- 1. Vinay Kumar, Abul K. Abas, Jon C. Aster; Robbins & Cotran Pathologic Basis of Disease; South Asia edition; India; Elsevier; 2014.
- 2. Harsh Mohan; Text book of Pathology; 6th edition; India; Jaypee Publications; 2010.
- 3. Laurence B, Bruce C, Bjorn K.; Goodman Gilman's The Pharmacological Basis of Therapeutics; 12th edition; New York; McGraw-Hill; 2011.
- 4. Best, Charles Herbert 1899-1978; Taylor, Norman Burke 1885-1972; West, John B (John Burnard); Best and Taylor's Physiological basis of medical practice; 12th ed; united states:
- 5. William and Wilkins, Baltimore; 1991 [1990 printing].
- 6. Nicki R. Colledge, Brian R. Walker, Stuart H. Ralston; Davidson's Principles and Practice of Medicine; 21st edition; London; ELBS/Churchill Livingstone; 2010.
- 7. Guyton A, John .E Hall; Textbook of Medical Physiology; 12th edition; WB Saunders Company; 2010.
- 8. Joseph DiPiro, Robert L. Talbert, Gary Yee, Barbara Wells, L. Michael Posey; Pharmacotherapy: A Pathophysiological Approach; 9th edition; London; McGraw-Hill Medical; 2014.
- 9. V. Kumar, R. S. Cotran and S. L. Robbins; Basic Pathology; 6th edition; Philadelphia; WB Saunders Company; 1997.
- 10. Roger Walker, Clive Edwards; Clinical Pharmacy and Therapeutics; 3rd edition; London; Churchill Livingstone publication; 2003.

Recommended Journals

- 1. The Journal of Pathology. ISSN: 1096-9896 (Online)
- 2. The American Journal of Pathology. ISSN: 0002-9440
- 3. Pathology. 1465-3931 (Online)
- 4. International Journal of Physiology, Pathophysiology and Pharmacology. ISSN: 1944-8171 (Online)
- 5. Indian Journal of Pathology and Microbiology. ISSN-0377-4929.



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: General

SEMESTER: II CODE: BP205T

NAME: Computer Applications in Pharmacy – Theory

Teaching & Evaluation Scheme:-

I caciii	ing & Dyuluu				,										
		Teac	hing S	cheme	e (Hours)					Evalua	ation Sche	me			
Subject	Name of the					Credits		The	eory			Prac	ctical		
Code	ode Subject Th Tu Pr T				Total		Intern		End Seme		Interi		End Sem		Total
							Exan	1	Exan	1	Exai	n	Exar	n	
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
	Computer	_			_	_	15	1							
BP205T	Applications in Pharmacy -Theory	3	0	0	3	3	10 (CM)		50	2					75

Scope: This subject deals with the introduction Database, Database Management system, and computer application in clinical studies and use of databases.

Objectives: Upon completion of the course the student shall be able to

- know the various types of application of computers in pharmacy
- know the various types of databases
- know the various applications of databases in pharmacy

COURSE CONTENT (30 Hours)

UNIT-I 06 Hours

- Number system: Binary number system, Decimal number system, Octal number system,
 Hexadecimal number systems, conversion decimal to binary, binary to decimal, octal to
 binary etc, binary addition, binary subtraction One's complement, Two's complement
 method, binary multiplication, binary division
- Concept of Information Systems and Software: Information gathering, requirement
 and feasibility analysis, data flow diagrams, process specifications, input/output design,
 process life cycle, planning and managing the project



UNIT-II 06 Hours

 Web technologies: Introduction to HTML, XML,CSS and Programming languages, introduction to web servers and Server Products
 Introduction to databases, MYSQL, MS ACCESS, Pharmacy Drug database

UNIT-III 06 Hours

UNIT-IV 06 Hours

• **Bioinformatics:** Introduction, Objective of Bioinformatics, Bioinformatics Databases, Concept of Bioinformatics, Impact of Bioinformatics in Vaccine Discovery

UNIT-V 06 Hours

• Computers as data analysis in Pre-clinical development: Chromatographic dada analysis(CDS), Laboratory Information management System (LIMS) and Text Information Management System(TIMS)



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: General

SEMESTER: II CODE: BP210P

NAME: Computer Applications in Pharmacy - Practical

Teaching & Evaluation Scheme:-

		Teac	hing S	chem	e (Hours)					Eval	uation Sc	heme			
Subject	Name of the					Credits		Th	eory			Prac	ctical		
Code	Code Subject		Tu	Pr	Total		Interi Exai		End Semo		Interi Exai		End Sem Exan		Total
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
	Computer										5	2			
BP210P	Applications in Pharmacy – Practical			2	2	1					5 (CM)		15	2	25

- 1. Design a questionnaire using a word processing package to gather information about a particular disease.
- 2. Create a HTML web page to show personal information.
- 3. Retrieve the information of a drug and its adverse effects using online tools
- 4. Creating mailing labels Using Label Wizard, generating label in MS WORD
- 5. Create a database in MS Access to store the patient information with the required fields Using access
- 6. Design a form in MS Access to view, add, delete and modify the patient record in the database
- 7. Generating report and printing the report from patient database
- 8. Creating invoice table using MS Access
- 9. Drug information storage and retrieval using MS Access
- 10. Creating and working with queries in MS Access
- 11. Exporting Tables, Queries, Forms and Reports to web pages
- 12. Exporting Tables, Queries, Forms and Reports to XML pages



Recommended books (Latest edition):

- 1. Computer Application in Pharmacy William E. Fassett –Lea and Febiger, 600 South Washington Square, USA, (215) 922-1330.
- 2. Computer Application in Pharmaceutical Research and Development –Sean Ekins Wiley-Interscience, A John Willey and Sons, INC., Publication, USA
- 3. Bioinformatics (Concept, Skills and Applications) S. C. Rastogi CBS Publishers and Distributors, 4596/1- A, 11 Darya Gani, New Delhi 110 002(INDIA)
- Microsoft office Access 2003, Application Development Using VBA, SQL Server, DAP and Infopath –Cary N. Prague – Wiley Dreamtech India (P) Ltd., 4435/7, Ansari Road, Daryagani, New Delhi - 110002



FACULTY OF: Pharmaceutical Sciences

DEPARTMENT OF: General

SEMESTER: II CODE: BP206T

NAME: Environmental Sciences - Theory

Teaching & Evaluation Scheme:-

		Teac	hing So	cheme	(Hours)					Evalua	tion Sche	me			
Subject	Name of the					Credits		The	eory			Prac	ctical		
Code	Code Subject Th Tu Pr Tota		Total		Intern Exar		End Seme Exam		Interi Exai		End Sem Exar		Total		
							Marks	Hrs	Marks	Hrs	Marks	Hrs	Marks	Hrs	
DD204T	Environmental	_			2	2	15	1	50	2					
BP206T	Sciences -Theory	3	0	0	3	3	10 (CM)		50	2					75

Scope: Environmental Sciences is the scientific study of the environmental system and the status of its inherent or induced changes on organisms. It includes not only the study of physical and biological characters of the environment but also the social and cultural factors and the impact of man on environment.

Objectives: Upon completion of the course the student shall be able to:

- Create the awareness about environmental problems among learners.
- Impart basic knowledge about the environment and its allied problems.
- Develop an attitude of concern for the environment.
- Motivate learner to participate in environment protection and environment improvement.
- Acquire skills to help the concerned individuals in identifying and solving environmental problems.
- Strive to attain harmony with Nature.

COURSE CONTENT (30 Hours)

UNIT-I 10 Hours

The Multidisciplinary nature of environmental studies

Natural Resources

Renewable and non-renewable resources: Natural resources and associated problems



a) Forest resources; b) Water resources; c) Mineral resources; d) Food resources; e) Energy resources; f) Land resources: Role of an individual in conservation of natural resources.

UNIT-II 10 Hours

Ecosystems

- Concept of an ecosystem.
- Structure and function of an ecosystem.
- Introduction, types, characteristic features, structure and function of the ecosystems: Forest ecosystem; Grassland ecosystem; Desert ecosystem; Aquatic ecosystems (ponds, streams, lakes, rivers, oceans, estuaries)

UNIT-III 10 Hours

Environmental Pollution: Air pollution; Water pollution; Soil pollution

Recommended Books (Latest edition):

- 1. Y.K. Sing, Environmental Science, New Age International Pvt, Publishers, Bangalore
- 2. Agarwal, K.C. 2001 Environmental Biology, Nidi Publ. Ltd. Bikaner.
- 3. Bharucha Erach, The Biodiversity of India, Mapin Pu blishing Pvt. Ltd., Ahmedabad 380 013, India,
- 4. Brunner R.C., 1989, Hazardous Waste Incineration, McGraw Hill Inc. 480p
- 5. Clark R.S., Marine Pollution, Clanderson Press Oxford
- 6. Cunningham, W.P. Cooper, T.H. Gorhani, E & Hepworth, M.T. 2001, Environmental Encyclopedia, Jaico Publ. House, Mumbai, 1196p
- 7. De A.K., Environmental Chemistry, Wiley Eastern Ltd.
- 8. Down of Earth, Centre for Science and Environment